

SYNTHESIS OF CHIRAL OLEFINS BY THE WITTIG REACTION.

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(Received in UK 12 October 1976; accepted for publication 29 November 1976)

The work made up to now on the stereochemistry of the Wittig reaction mainly deals with factors influencing the geometrical isomerism of the produced internal olefins ¹ or with the retention of absolute configuration at the phosphorus atom when chiral phosphines are employed ². There are no exhaustive reports on the stereospecificity of this reaction with respect to the optical isomerism in preparing chiral olefins with an asymmetric carbon atom α or β to the double bond.

The series of internal cis and trans olefins reported in Table 1, characterized by the presence of one or two asymmetric carbon atoms with high prevalence of S chirality, was prepared following the Hauser procedure for the Wittig reaction ³. With the sole exception of (S)-4-methylhex-2-ene ⁴, III, these olefins have not been described till now.

According to the olefin structure, the chiral aldehydes ^{5,6} and the chiral bromides ⁷ were both necessarily used in the preparation of (3S,6S)-3,6-dimethyloct-4-ene, I, and (3S,8S)-3,8-dimethyldec-5-ene, IV. (S)-2,6-dimethyloct-4-ene, II, and III were both obtained either from (S)-2-methylbutanal and the suitable bromide or from (S)-1-bromo-2-methylbutane and the suitable aldehyde. Finally (S)-2,7-dimethylnon-4-ene, V, was prepared by using the chiral aldehyde (S)-3-methylpentanale ⁶ and 1-bromo-3-methylbutane.

The yield strongly depends on the structure of the reagents ³, and the cis/trans ratio is as high as expected in the case of unstabilized ylids ¹ (Table 1). The separation of the geometrical isomers to a purity greater than 99.5% was successfully achieved by a Perkin Elmer Model 251 spinning band column or by preparative gas-chromatography using AgNO₃ in ethyleneglycol as a stationary phase.

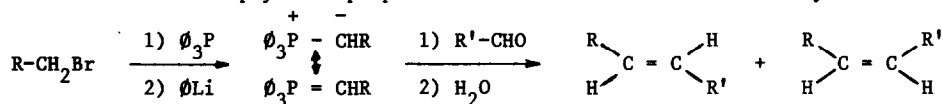
The structure of the olefins prepared and some physical properties of their geometrical isomers are reported in Tables 1 and 2.

The optical purity of the samples was determined by converting the olefin into the corresponding ozonide (CHCl₃, -30°) and successive reduction to the alcohol by LiAlH₄ (Scheme, route A).

In all cases no racemization was observed by comparing the optical purity of the alcohol with that of the bromide or of the aldehyde employed in the synthesis of the olefin, so that the same optical purity of the alcohol can be assigned to the latter (Table 2).

Table 1

Structure and some physical properties of chiral olefins obtained by the Wittig reaction



Olefin	R	R'	Yield ^a (%)	cis/trans ^b	Geom. isomer	b.p. (°C)	d ₄ ²⁵	n _D ²⁵
I	$\begin{matrix} \text{C}_2\text{H}_5-\text{CH}- \\ \\ \text{CH}_3 \end{matrix}$	$\begin{matrix} \text{C}_2\text{H}_5-\text{CH}- \\ \\ \text{CH}_3 \end{matrix}$	28	67/33	cis	149-151	0.7316	1.4166
					trans	154-155	0.7302	1.4166
II	$\begin{matrix} \text{CH}_3-\text{CH}-\text{CH}_2- \\ \\ \text{CH}_3 \end{matrix}$	$\begin{matrix} \text{C}_2\text{H}_5-\text{CH}- \\ \\ \text{CH}_3 \end{matrix}$	60	70/30	cis	154-155	0.7359	1.4184
					trans	156-157	0.7351	1.4164
III	$\begin{matrix} \text{CH}_3- \\ \\ \text{C}_2\text{H}_5-\text{CH}- \\ \\ \text{CH}_3 \end{matrix}$	$\begin{matrix} \text{C}_2\text{H}_5-\text{CH}- \\ \\ \text{CH}_3 \end{matrix}$	58	79/21	cis ^c	87-88	0.6951	1.4001
					trans ^c	87-88	0.6932	1.3997
IV	$\begin{matrix} \text{C}_2\text{H}_5-\text{CH}-\text{CH}_2- \\ \\ \text{CH}_3 \end{matrix}$	$\begin{matrix} \text{C}_2\text{H}_5-\text{CH}-\text{CH}_2- \\ \\ \text{CH}_3 \end{matrix}$	78	68/32	cis	93-94 ^d	0.7676	1.4365
					trans	93-94 ^d	0.7622	1.4311
V	$\begin{matrix} \text{CH}_3-\text{CH}-\text{CH}_2- \\ \\ \text{CH}_3 \end{matrix}$	$\begin{matrix} \text{C}_2\text{H}_5-\text{CH}-\text{CH}_2- \\ \\ \text{CH}_3 \end{matrix}$	75	70/30	cis	83-84 ^e	0.7529	1.4278
					trans	83-84 ^e	0.7464	1.4244

^a computed with respect to the phosphonium salt^b determined by vapour phase chromatography on the reaction product^c for the physical constants reported in literature see reference 4^d measured at 16 mm Hg^e measured at 20 mm Hg

Table 2

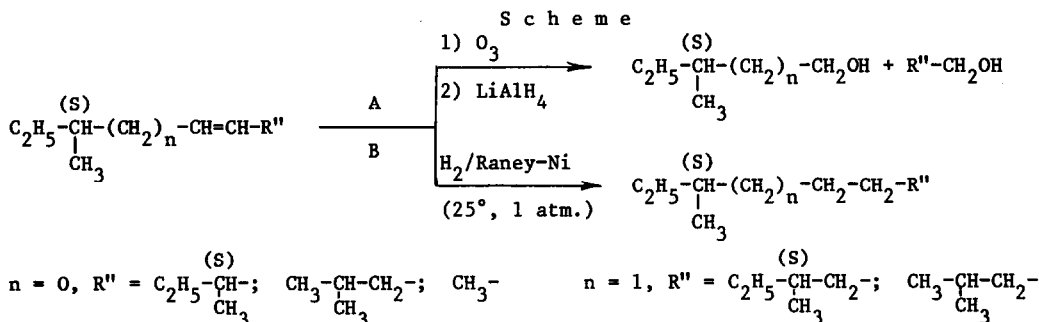
Relationship between optical rotatory power and optical purity (o.p.) of the olefins and racemization in their hydrogenation.

Chiral olefin		Chiral compound used in the synthesis		Alcohol (scheme, route A)		Paraffin (scheme, route B)	
Structure	Isomer $[\alpha]_D^{25}$	Type	o.p. (%)	$[\alpha]_D^{25}$	o.p. (%)	$[\alpha]_D^{25}$	o.p. (%)
							Racemization (%)
I $C_2H_5-\overset{\text{H}}{\underset{\text{CH}_3}{\text{C}}}-\overset{\text{H}}{\underset{\text{H}}{\text{C}}}=\overset{\text{H}}{\underset{\text{H}}{\text{C}}}-\overset{\text{H}}{\underset{\text{H}}{\text{C}}}-C_2H_5$	cis +50.67	aldehyde	96.0	-5.59	96.1 ^a	+14.96	71 ^b
	trans +64.83	bromide	97.0	-5.60	95.9 ^a		25
II $C_2H_5-\overset{\text{H}}{\underset{\text{CH}_3}{\text{C}}}-\overset{\text{H}}{\underset{\text{H}}{\text{C}}}=\overset{\text{H}}{\underset{\text{H}}{\text{C}}}-\overset{\text{H}}{\underset{\text{H}}{\text{C}}}-CH_3$	cis +17.41	aldehyde	96.0	-	-	+9.0	89 ^c
	trans +29.19						7
III $C_2H_5-\overset{\text{H}}{\underset{\text{CH}_3}{\text{C}}}-\overset{\text{H}}{\underset{\text{H}}{\text{C}}}=\overset{\text{H}}{\underset{\text{H}}{\text{C}}}-CH_3$	cis +28.57	aldehyde	96.0	-5.56	95.6 ^a	+8.73	88 ^d
	trans +37.64						8
III $C_2H_5-\overset{\text{H}}{\underset{\text{CH}_3}{\text{C}}}-\overset{\text{H}}{\underset{\text{H}}{\text{C}}}=\overset{\text{H}}{\underset{\text{H}}{\text{C}}}-CH_3$	cis +28.97	bromide	97.5	-5.66	97.3 ^a	-	-
	trans +38.09						-
IV $C_2H_5-\overset{\text{H}}{\underset{\text{CH}_3}{\text{C}}}-\overset{\text{H}}{\underset{\text{H}}{\text{C}}}-CH_2-\overset{\text{H}}{\underset{\text{CH}_3}{\text{C}}}-C_2H_5$	cis +7.61	aldehyde	96.4	+8.40	96.0 ^b	+16.1	92 ^b
	trans +3.98	bromide	96.4				4
V $C_2H_5-\overset{\text{H}}{\underset{\text{CH}_3}{\text{C}}}-\overset{\text{H}}{\underset{\text{H}}{\text{C}}}-CH_2-\overset{\text{H}}{\underset{\text{CH}_3}{\text{C}}}-CH_3$	cis +4.04	aldehyde	96.4	+8.39	96.0 ^b	-	-
	trans +2.56						-

^a L.Lardicci and P.Pino, Gazz.Chim.Ital., 91, 441 (1961).^b P.Pino, L.Lardicci, and A.Stefani, Ann.Chim., 52, 456 (1962).^c F.Ciardelli and P.Pino, Gazz.Chim.Ital., 95, 1201 (1965).^d P.A.Levane and R.E.Marker, J.Biol.Chem. 103, 302 (1933).

From the results obtained one has to conclude that the Wittig reaction occurs without racemization even if a tertiary asymmetric carbon atom is in the α position to the double bond, both starting with the bromide or the aldehyde as optically active materials.

This fact makes the Wittig reaction a very useful stereospecific method for the synthesis of chiral olefins in high optical yield.



Following the route B (Scheme), that in principle could be an alternative method for the optical purity determination, a racemization of 4 to 25% was detected by comparing the optical purity of the paraffins with that of the chiral starting compounds.

Such a racemization cannot be connected with the Wittig reaction as shown by the reductive ozonization of the olefin and hence it occurs during the hydrogenation of the double bond.

In this connection it is worth noting that, contrary to what observed in the case of the α -olefins^{8,9}, the catalytic hydrogenation of chiral olefins with an internal double bond produces racemization in some extent.

R e f e r e n c e s

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